

### Michael Caris (Autor) A Supersonic Jet Spectrometer for Terahertz Applications (SuJeSTA)



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## Introduction

The interstellar medium (ISM) contains a rich variety of molecular species of which more than 130 have been identified beyond doubt by means of their characteristic spectra. Organic and inorganic species with up to 13 atoms have been found in interstellar molecular clouds, circumstellar regions, gas outflows, and planetary atmospheres (see Tab. 2.1). So far, the heaviest element in any detected interstellar compound is iron which is the ninth abundant atom. The elements with masses up to that of Fe are produced in nuclear synthesis of active stars and amount almost 100 % of the cosmic mass. The heavier, less abundant (0.0005 %) elements are generated in super novae by proton- or neutron-accretion. The 22 most abundant elements are listed in Tab. 2.2.



Figure 2.1: Schematic diagram of the electromagnetic spectrum and corresponding interactions of radiation with molecules.

Information about interstellar objects is mainly obtained via detecting electromagnetic radiation which is either emitted or absorbed by gas phase molecules, atoms, and ions. Due to the large temperature scale, ranging from a few Kelvin, in dense molecular clouds and star forming regions, to several hundred Kelvin, in circumstellar shells, a great part of the electromagnetic spectrum is appropriate for analyzing interstellar objects. It is the submillimeter and millimeter wavelength region (far-infrared), where molecular rotation transitions dominate, adjacent the near- and mid-infrared regime, with vibrations, followed by the optical and ultraviolet part of the spectrum which is caused by electronic transitions (see Fig. 2.1). The unambiguous identification of an interstellar molecular species strongly depends on the precise knowledge of its spectrum which is obtained by measuring transition frequencies in the laboratory and by deriving molecular parameters for further line predictions. Powerful tunable radiation sources, for instance synthesizers and Backward Wave Oscillators (BWOs), are used in laboratory spectrometers to generate radiation in the frequency region up to 1 THz. Especially the spectra of the centimeter and millimeter wavelength region are of great importance, since most of the molecules have been detected in the frequency region between 10 and 300 GHz (see Tab. 2.1).

At frequencies above 500 GHz the earth's atmosphere starts getting opaque for electromagnetic radiation and radioastronomic observations are hampered mainly due to atmospheric water. In near future, telescopes based on new receiver techniques, such as APEX<sup>1</sup>, the airborne observatory SOFIA<sup>2</sup>, and the HIFI<sup>3</sup> instrument aboard the Herschel satellite, will extend the observable spectral range towards shorter wavelengths, opening the terahertz domain for radioastronomy. In return, new techniques, such as superlattice devices combined with BWOs [24] or laser sideband generation [25], provide electromagnetic radiation sources above 1 THz for laboratory applications (see Fig. 2.2).



Figure 2.2: Schematic diagram of the frequency coverage of the Cologne terahertz radiation sources (two lower rows) compared to the frequency range of some telescopes (top rows).

In general, molecules split into two groups: Molecules which have a permanent dipole moment and those which do have not. The nonpolar species, for instance  $C_n$ ,  $CO_2$ ,  $C_2H_2$ , or  $CH_4$ , have no pure rotational spectrum. They

<sup>&</sup>lt;sup>1</sup>Atacama Pathfinder Experiment, 200  $\mu m$  – 230  $\mu m$  and 300  $\mu m$  – 1500  $\mu m$ 

<sup>&</sup>lt;sup>2</sup>Stratospheric Observatory For Infrared Astronomy, 5  $\mu m$  – 300  $\mu m$ 

<sup>&</sup>lt;sup>3</sup>Heterodyne Instrument for the Far Infrared, 156  $\mu m$  – 212  $\mu m$  and 240  $\mu m$  – 624  $\mu m$ 

2 atoms	3 atoms	4 atoms	5 atoms	6 atoms
H2	$C_3^*$	$c - C_3 H$	<i>C</i> 5*	C <sub>5</sub> H
AlF	$C_2H$	$l - C_3 H$	$C_{\Delta}H$	$l - C_2 H_4$
AlCl	$\tilde{C_2O}$	$C_3N$	$C_4Si$	$C_2H_4^*$
$C_{2}^{**}$	$\tilde{C_2S}$	$C_3O$	$l-C_3H_2$	$CH_3CN$
ĊH	$\overline{CH_2}$	$C_3S$	$c - C_3 H_2$	$CH_3NC$
$CH^+$	HCN	$C_{2}H_{2}^{*}$	$CH_2CN$	CH <sub>3</sub> OH
CN	НСО	NH <sub>3</sub>	$CH_4^*$	$CH_3SH$
СО	$HCO^+$	HCCN	$HC_3N$	$HC_3NH^+$
$CO^+$	$HCS^+$	$HCNH^+$	$HC_2NC$	$HC_2CHO$
СР	$OHC^+$	HNCO	HCOOH	NH <sub>2</sub> CHO
SiC	$H_2O$	HNCS	$H_2CNH$	$C_5N$
HCl	$H_2S$	$HOCO^+$	$H_2C_2O$	$l - HC_4 H^*$ (?)
KCl	HNC	$H_2CO$	$H_2NCN$	$l - HC_4N$
NH	HNO	$H_2CN$	$HNC_3$	
NO	MgCN	$H_2CS$	$SiH_4^*$	
NS	MgNC	$H_3O^+$	$H_2COH^+$	
NaCl	$N_2H^+$	$c-SiC_3$		
ОН	$N_2O$	$CH_3^*$		
PN	NaCN			
SO	OCS			
$SO^+$	$SO_2$			
SiN	$c-SiC_2$			
SiO	$CO_2^*$			
SiS	$NH_2$			
CS	$H_{3}^{+*}$			
HF	$H_2D^+$			
$SH^*$	$HD_2^+$			
HD	SiCN			
<i>FeO</i> (?)	AlNC			
$O_2$ ?	SiNC			
7 atoms	8 atoms	9 atoms	10 atoms	11–13 atoms
C <sub>6</sub> H	CH <sub>3</sub> C <sub>3</sub> N	$CH_3C_4H$	CH <sub>3</sub> C <sub>5</sub> N	HC <sub>9</sub> N
$CH_2CHCN$	HCOOCH <sub>3</sub>	$CH_3CH_2CN$	$(CH_3)_2CO$	$C_{6}H_{6}^{*}(?)$
$CH_3C_2H$	CH <sub>3</sub> COOH	$(CH_3)_2O$	$(CH_2OH)_2$ (?)	$CH_3OC_2H_5$
$HC_5N$	$C_7H$	$CH_3CH_2OH$	$H_2NCH_2COOH$ ?	$HC_{11}N$
CH <sub>3</sub> CHO	$H_2C_6$	$HC_7N$	CH <sub>3</sub> CH <sub>2</sub> CHO	
$CH_3NH_2$	$CH_2OHCHO$	$C_8H$		
$c - C_2 H_4 O$	$l - HC_6 H^*$ (?)			
H <sub>2</sub> CCHOH	$CH_2CHCHO(?)$			

Table 2.1: List of interstellar molecules detected by means of their rotational spectra in the radio to far-infrared frequency region.

\* indicates molecules that have been detected by ro-vibrational spectroscopy.

\*\* indicates molecules that have been detected by electronic spectroscopy.

l indicates linear molecules.

c indicates cyclic molecules.

? indicates a questionable or (probable) tentative detection

can be identified by means of infrared active transitions in the near- and midinfrared belonging to ro-vibrational modes. The molecules with a permanent dipole moment can be traced by means of pure rotational transitions in the millimeter and submillimeter wavelength region. Additionally, many chain molecules also have low energy bending vibrations in the terahertz region.

An important class of interstellar species are diatomic compounds. The list of diatomic molecules detected currently comprises of 30 entries two third of which are inorganic species. In 1987, Cernicharo and Guélin reported the detection of four metal halides, i.e. *KCl*, *NaCl*, *AlF*, and *AlCl*, towards the circumstellar envelope of late type star IRC+10216 [14]. The relative high cosmic abundance of aluminium, sodium, potassium, and chlorine, i.e. the  $13^{th}$ ,  $14^{th}$ ,  $20^{th}$ , and  $19^{th}$  most abundant element (see Tab. 2.2), lend encouraging motivation to continue searching for metal halides towards hotter and even denser core regions. The recently detected rotational lines of *NaCl* on Jupiter's moon Io at frequencies of 143 and 234 *GHz* [15] demonstrate how special conditions, for instance continuous volcanic output, can produce spectroscopical measurable amounts of *KCl*, *NaCl*, etc.

Experimental studies on diatomic gas phase alkali halides in the laboratory are complicated due to the low vapor pressure of these molecules. Special vaporization conditions are necessary to provide them for gas phase spectroscopy. On the other hand, these species possess a very large dipole moment because of their highly ionic character. This means, they reveal a pure rotational spectrum with mainly large intensities. Reliable data on the alkali halides is rare. First spectroscopic data of NaCl, RbCl, and CsCl in the millimeter wavelength range was published already in 1964 [3]. In 1997, the infrared emission spectra of KCl and NaCl were published [16]. First high resolution measurements on sodium chloride in the THz region were performed with the Cologne terahertz spectrometer in 2001 (Caris *et al.* [26], [2]).

In the course of this work, high resolution measurements on five isotopomers of potassium chloride, i.e.  ${}^{39}K^{35}Cl$ ,  ${}^{39}K^{37}Cl$ ,  ${}^{41}K^{35}Cl$ ,  ${}^{41}K^{37}Cl$  and  ${}^{40}K^{35}Cl$ , have been performed up to 930 *GHz* with the Cologne terahertz spectrometer. A special evaporation cell has been used. The obtained data has been analyzed in an isotopically invariant form and a new set of invariant Dunham parameters has been derived including the first order Born-Oppenheimer breakdown corrections. With these molecular constants, reliable frequency predictions for the spectra of all six isotopomers are available which open the field for further astronomical detections.

A plethora of detected interstellar molecules contains one or more carbon atoms (75 %). Carbon is the fourth abundant element, after hydrogen, helium, and oxygen, and reveals a wealth of possible chemical bondings. One characteristic feature of the carbon atom is the tendency to build chain molecules, such as the stable cyano-polyynes,  $HC_nN$  (with n = 1, 3, 5, ...). This type of molecules has been detected in many astronomical sources, for instance towards the circumstellar envelope IRC+10216 (e.g. HCN [28], [29];  $HC_3N$ 

Element	Symbol	Abundance
		10
Hydrogen	Н	$3.18 \cdot 10^{10}$
Helium	He	$2.21 \cdot 10^{9}$
Oxygen	0	$2.15 \cdot 10^{7}$
Carbon	С	$1.18 \cdot 10^{7}$
Nitrogen	Ν	$3.74 \cdot 10^{6}$
Neon	Ne	$3.44 \cdot 10^{6}$
Magnesium	Mg	$1.061 \cdot 10^{6}$
Silicon	Si	$1.00\cdot 10^6$
Iron	Fe	$8.3 \cdot 10^5$
Sulfur	S	$5.0 \cdot 10^{5}$
Argon	Ar	$1.172 \cdot 10^5$
Calcium	Ca	$7.21 \cdot 10^4$
Aluminium	Al	$8.5 \cdot 10^4$
Sodium	Na	$6.0 \cdot 10^4$
Nickel	Ni	$4.80 \cdot 10^4$
Chromium	Cr	$1.27\cdot 10^4$
Phosphorus	Р	9600
Manganese	Mn	9300
Chlorine	Cl	5700
Potassium	Κ	4200
Titanium	Ti	2775
Fluorine	F	2450
All other elements		5069

Table 2.2: Cosmic abundance of the elements normalized to  $Si = 10^6$ . Values taken from the publication of A. G. W. Cameron [27].

#### $[30]; HC_5N, HC_7N [31]).$

Besides the persistent carbon chains, there are numerous unsaturated and reactive molecules, for instance  $C_nH$ ,  $C_nN$ , pure carbon chains  $C_n$  (with n = 1, 2, 3, ..., and those cyano-polyynes with even numbers of carbon atoms. They are of significant astrophysical relevance and have been observed, for instance towards cold molecular clouds, such as TMC-1 and L183 (e.g.  $l-C_3H$ ,  $C_4H, C_5H, C_6H$  [5]). Unlike the laboratory, the interstellar medium with ultralow pressure, density, and temperature conditions is a repository for this class of molecules. Thus, many radicals have been detected first by radio astronomers rather than in the laboratories, for example  $C_3N$  [32] and  $C_4H$ ([33], [34], and [35]) which were first observed in 1977 and 1978, respectively. The frequency predictions for theses detections had to be taken from ab inito calculations, since laboratory spectroscopy could not catch up with the interstellar detections for a long time until sophisticated production methods were developed. Nowadays, laboratory spectrometers for the investigation of radicals are available mainly in the infrared region, while the terahertz range is barely involved in the investigation of reactive species.

The Cologne Supersonic Jet Spectrometer for Terahertz Applications (Su-JeSTA) has been developed within the scope of this thesis to overcome this lack of terahertz data for radicals. The new instrument combines an efficient source for radicals and a powerful terahertz radiation source to obtain high resolution spectra of the instable species and to provide the necessary laboratory data for future search for reactive, interstellar species. The molecules are formed by means of a pulsed discharge in the throat of a slit nozzle. The subsequent adiabatic, supersonic expansion into the vacuum chamber is almost collision-free and cools the rotational degree of freedom of the gas molecules to a few Kelvin. SuJeSTA is embedded in the international network of the Laboratoire Europeén Associé (LEA *HiRes*) which aims at developing a regional structure promoting the applications of high resolution spectroscopy in the field of molecular physics. Applications concern planetary science, atmospheric studies, astrophysics of the interstellar medium, time dependent chemistry, analytical chemistry, and process analysis.

The first application of SuJeSTA has been an investigation of the linear  $X^2\Pi$  propynylidyne radical,  $l-C_3H$ . This chain molecule has been subject of many astronomical observations, since it is supposed to play a major role in the carbon chain growth in the interstellar medium. The linear  $C_3H$  radical in its vibrational ground state was first detected by Thaddeus *et al.* [4] towards IRC+10216 and the cold dark cloud TMC-1 in 1985. In 2000, Turner *et al.* [5] reported cyclic and linear  $C_3H$  in three small translucent molecular clouds (CB 17, CB 24, and CB 228), in TMC-1, and L183. The most recent detection of  $C_3H$  stems from an extensive spectral line survey towards TMC-1 (Kaifu *et al.* in 2004 [6]).

The  $C_3H$  radical is not only of astrophysical interest, but also a challenging molecule for spectroscopists. First information on  $C_3H$  came from theoretical studies. *Ab initio* calculations revealed the equilibrium geometric structure and provided vibrational frequencies ([17], [18], [19]). Effects, such as spin-orbit coupling due to the  ${}^{2}\Pi$  electronic ground state, and the hyperfine structure due to the nuclear spin of the hydrogen, were treated in many publications (e.g. [36], [37], [20]). Recently Perić *et al.* [21] investigated the Renner-Teller effect of  $C_{3}H$ , resulting from a coupling of the electronic angular momentum and the bending vibrational motion. Due to this strong coupling, the lower  ${}^{2}\Sigma^{\mu}$ -state of the  $v_{4}$  *CCH* bending mode is shifted towards lower energies and into the submillimeter region. This effect is typical for carbon chains and the understanding of its nature will support *ab initio* calculations of other chain molecules.

SuJeSTA has facilitated the first experimental study of this Renner-Teller shift due to the precise data of ro-vibrational transitions of the  ${}^{2}\Sigma^{\mu} {}^{2}\Pi_{3/2}$  band system, which is described in this work. A reasonable value for the Renner-Teller constant  $\varepsilon$  of  $C_{3}H$  has been evaluated, based on these measurements. Furthermore, the value of the excitation energy  $E_{\nu_{4}}$  of the  $\nu_{4}$  ( ${}^{2}\Sigma^{\mu}$ ) state has been strongly improved.

Another interesting feature of  $C_3H$  is a Coriolis coupling which causes perturbations of the rotational levels in the ground and the vibrationally excited states. First accurate values for the Coriolis interaction constant will be presented in this work. The pure rotational transitions in both  ${}^2\Pi$  ground states and in the first excited vibrational state  $v_4 = 1$  ( ${}^2\Sigma^{\mu}$ ) have been recorded up to 600 *GHz* and the earlier measurements by Gottlieb *et al.* (1985 [7] and 1986 [8]) and Yamamoto *et al.* (1990 [9]) below 360 *GHz* have been substantially extended. From a least squares fit to a standard Hamiltonian a new set of molecular parameters has been obtained which is most reliable to predict transition frequencies up to 1 *THz*.

Besides uncharged molecules, ions represent another class of astrophysically important species. Like radicals, these molecules are usually instable under laboratory conditions and difficult to analyze by spectroscopical methods. However, they are abundant in the interstellar medium and play a major role in astrochemistry, since many chemical reactions involve ions.

Furthermore, ions, such as  $CO^+$  and  $C^+$ , play a decisive role as PDR tracer (Photon Dominated Region). In the interstellar medium, the  $CO^+$  ion is rare and normally optically thin due to its reactivity, for instance with  $H_2$ to  $HCO^+$ . With knowledge of the column density of  $CO^+$  chemical models of PDRs are verified [11]. The first detection of  $CO^+$  towards OMC-1, a molecular cloud in the Orion nebula, was reported by Erickson *et al.* [12] in 1981. Twelve years later, Latter *et al.* [38] observed  $CO^+$  in the planetary nebula NGC7027 and in the interstellar medium (M17SW).

SuJeSTA facilitates the analysis of adiabatically cooled ions in the terahertz region - in contrast to the production of ions in a DC glow discharge, where the ions usually have high rotational temperatures [10]. In this work, the carbon monoxide ion,  $CO^+$ , has been used as a test molecule. It turned out, that SuJeSTA is an expedient instrument for the spectroscopy of cold ions in the terahertz region and pure rotational transitions of  $CO^+$  in the vibrational ground state have been observed.

 $CO^+$ , generated in glow discharges, was investigated in many laboratory mi-

crowave studies in the 1970s and 1980s ([39], [22], [40], and [23]). The latest set of isotopically invariant Dunham parameters was determined in the early 1980s. A very recent extensive investigation of the isotopomers  ${}^{12}CO^{+}$  and  ${}^{13}CO^{+}$  in the terahertz domain was carried out by Klapper [10] employing the Cologne Terahertz Spectrometer.

An updated isotopically invariant analysis of  $CO^+$  including the Born-Oppenheimer breakdown corrections was not available, yet. Thus, the observed lines together with earlier data have been fitted in a mass independent least squares fit. The isotopically invariant mass-reduced Dunham parameters and the first order Born-Oppenheimer breakdown corrections for C and O have been determined. The resultant invariant parameter set of this work provides reliable frequency predictions for all isotopomers of  $CO^+$  and encourages the astronomical search for those with low abundances.

Although the species which have been analyzed in this thesis require certain experimental setups and belong to different categories of molecules, they have all been investigated in the terahertz regime by means of laboratory spectroscopy. The rotational and, in case of  $C_3H$ , bending vibrational spectra have been analyzed with quantum mechanical methods to derive appropriate molecular parameters for their description. Upon these results the spectra are predictable in a wide range and the transition frequencies are available via the Cologne Database for Molecular Spectroscopy (CDMS) [41], [13]. 3

## **Experimental Setup**

The main aim of the present work is to combine a terahertz radiation source with a supersonic jet, to perform high resolution terahertz spectroscopy on molecular radicals and ions. For this purpose, the new Cologne Supersonic Jet Spectrometer for Terahertz Applications (SuJeSTA) has been setup which is described in Section 3.1. For the measurements on KCl the Cologne Terahertz Spectrometer has been used which has been described by Winnewisser *et al.* [1] in detail and in Section 3.2 of this work only briefly. Further details on the radiation sources are given in Section 3.3.

# 3.1 Supersonic Jet Spectrometer for Terahertz Applications

A main task in the course of this work was the construction and assembly of the new Cologne Supersonic Jet Spectrometer for Terahertz Applications (SuJeSTA). Extensive tests were performed to derive technical parameters, such as the amount of produced molecular transient species and lower limits for their detection. In a first scientific application SuJeSTA has been used to study pure rotational and ro-vibrational transitions of the linear  $C_3H$ radical (see Chapter 6). The new spectrometer has also been used to perform measurements on adiabatically cooled  $CO^+$  ions (see Chapter 7).

The short absorption pathes and the low column densities of a jet spectrometer require a strong and stable radiation source besides a very sensitive detecting system. Backward wave oscillators provide enough power in the submillimeter and millimeter wavelength region to operate a terahertz jet spectrometer. SuJeSTA can be used in two different operational modes (see Section 3.3.2), either in a frequency stabilized mode or in a free-running mode: A highly precise mode for scans over small intervals, usually of 10– 20 *MHz* width (see Fig. 3.1) and a fast scanning mode which covers a large frequency range of several hundred *MHz* (see Fig. 3.2). The less accurate